

#### Material Characterization for Processing: Hexcel 8552

Material Model Development Final Project Wrap-Up

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# **Project Definition**

Material: Hexcel 8552

Modeling performed for:

- Cure Kinetics
- Heat Capacity (Cp)
- Viscosity



#### **Material Description**

#### • HEXPLY 8552

- Material forms received from Hexcel are as follows:
  - Neat Resin: HS-AD-693 (received Oct 18, 2006)
  - Resin Film: 74#CCA1030/B430, 35G, 45.75 (received Oct 18, 2006)





# **Cure Kinetics Model**





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#### Nomenclature

Iso			
Int			
Dyn			
RES			









#### **HR Definition**







# **DSC Test Procedure**

#### Isothermal Tests

In isothermal tests, the sample is equilibrated at a very low temperature and then heated up to a predefined hold temperature at a very high rate. The sample is held at this temperature for a predetermined duration of time (i.e. until the reaction stops due to diffusion or full cure in the case of conditioned isothermal tests, or earlier as predefined for interrupted isothermal tests). The sample is then cooled down, followed by a residual ramp at a known rate (typically 1-4 cpm). The residual ramp ensures that the material is fully cured, and provides the material Tg at the end of hold, as well as the residual heat of reaction. A second ramp is also performed to determine the final (full cure) Tg.

#### Dynamic Tests

In dynamic tests, the sample is equili





# **DSC Tests Performed**

The DSC tests were performed are summarized below:

- Dynamic tests: 18 tests @ 1 to 10°Cpm
- Isothermal tests: 11 tests 100°C to 190°C
- Interrupted Isothermal tests: 2 tests at 160°C
- Other: 4 cure cycle tests, 4 dynamic on other forms

After careful examination of the results and investigation of the consistency of the data, the comp



#### **DSC** Tests Used in Model Fitting

Tost	Mass	Rate	<b>Temp</b> <sub>max</sub>
Test	(mg)	(ºC/min)	(°C)
8552-MDYN-01cpm-01	4.99	1	300
8552-MDYN-02cpm-01	4.19	2	300
8552-MDYN-02cpm-02	5.3	2	280
8552-MDYN-03cpm-01	5.59	3	300
8552-MDYN-04cpm-01	5.59	4	300
8552-MDYN-05cpm-01	4.79	5	300
8552-MDYN-06cpm-01	4.79	6	300
8552-MDYN-07cpm-01	5.89	7	300
8552-MDYN-08cpm-01	4.09	8	330
8552-MDYN-09cpm-01	5.19	9	330
8552-MDYN-10cpm-01	3.39	10	330

Mass (mg)





#### **Degree of Cure Calculation**

Raw DSC data was linearly sparsed and smoothed.

A bi-linear baseline was considered for dynamic tests in order to calculate the total heat of reaction and degree of cure. Baselines fitted to



# Cure Rate vs. DoC – Dyns



#### Cure Rate vs. DoC – Isos



# Heat Flow Response – 1cpm







# DoC and Cure Rate – 1cpm







#### Heat Flow Response – 2cpm-1







# DoC and Cure Rate – 2cpm-1







# Heat Flow Response – 3cpm







# DoC and Cure Rate – 3cpm







# DoC and Cure Rate – 4cpm







# Heat Flow Response – 5cpm







# DoC and Cure Rate – 5cpm







#### Heat Flow Response – 6cpm







27





#### Heat Flow Response – 7cpm







29

# Heat Flow Response – 9cpm







# DoC and Cure Rate – 9cpm







34

# DoC and Cure Rate – 10cpm










# Heat Flow Response – 120°C









# Heat Flow Response – 140°C







# DoC and Cure Rate – 140°C







# Heat Flow Response – 150°C







# Heat Flow Response – 160°C







### DoC and Cure Rate – 160°C











### DoC and Cure Rate – 180°C-2







# DoC and Cure Rate – 190°C







# **Total Heat of Reaction - Dyns**



# Total Heat of Reaction - Isos













### **Cure Kinetics Model**

1

**Chemical Reaction:** 

*k* 2

where





### **Cure Kinetics Model**

**Diffusion Component:** 

$$\dot{x}_{d} = K_{d0} e^{\frac{-B}{f}} F_{d}(x) \qquad \longrightarrow \qquad \dot{x}_{d} = 4.0 \times e^{\frac{-0.21}{f}}$$

where<sup>1</sup>  $f = a(T - T_g) + b$ 

with 
$$a = 4.8 \times 10^{-4}$$

and 
$$0.021$$
  $T_g < 120^{\circ}C$   
 $b = \text{Linear bw } 0.021 \text{ and } 0.031$   $120^{\circ}C < T_g < 195^{\circ}C$   
 $0.031$   $T_g > 195^{\circ}C$ 





### **Cure Kinetics Model**

$$\dot{x} = \frac{1}{\dot{x}_k} + \frac{1}{\dot{x}_d}^{-1}$$









### Ln(xdot)-1/T



Model-Kinetic4

# Ln(xdot)-1/T






















#### **Model Parameters**

Parameter	Reaction 1	Reaction 2
(1/s)	153,900.5	3.963E+11
(J/mol)	64,929.5	133,168.3
	2.347	1.029
	1.00	1.00
	0.00	0.00
	1.00	1.00
	0.1594	0.00
	1.413	5.586

Parameter	Value (unit)
	-7 (°C)
	250 (°C)
	0.78

Parameter	Value (unit)
	4.0 (1/s)
	0.21
	4.8E-04 (1/ºC)
	4.8E-04 (1/ºC)
	0 (°C)
	100 (°C)
	0.021
	0.031
	120 (°C)
	195 (ºC)



# Dynamic Tests – 1cpm





## Dynamic Tests – 2cpm-2







# Dynamic Tests – 3cpm







# Dynamic Tests – 4cpm







# Dynamic Tests – 8cpm







# Dynamic Tests – 9cpm











### Isothermal Tests - 100°C







### Isothermal Tests - 120°C







### Isothermal Tests - 130°C







### Isothermal Tests - 140°C







### Isothermal Tests - 150°C







### Isothermal Tests - 170°C







### Isothermal Tests - 180°C-1







### Isothermal Tests - 180°C-2







### Isothermal Tests - 190°C
































































## Isothermal Tests – All







# Post-Hold Tg





# Post-Hold Tg

Tg values at various points along the MRCC were measured and then compared to the predictions of the cure kinetics model. Good agreement was observed.







# Goodness of Fit - Overview

 Goodness of fit is measured by comparing the test to the model prediction at several key points in the







# Goodness of Fit – Peak





120



# Goodness of Fit – Other











## Goodnes s of Fit – Iso DOC







# Goodness of Fit – Dyn Timing







# Goodness of Fit – Dyn DOC







### **Other Forms - Fabric**





### **Other Forms - Fabric**











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## Other Forms – HR Values





#### Material Model Verification – 100C











#### Material Model Verification – 1Cpm

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1°C/min







140

3°C/min







14Z

4°C/min







5°C/min







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# Viscosity Model





## Nomenclature

- $\eta'$  Dynamic viscosity or the real part of the complex viscosity
- $\eta''$  Elastic complex part of the complex viscosity
- $\eta^*$  Complex viscosity
- $\mu$  Material viscosity calculated by the viscosity model





## Viscosity Tests

- Material viscosity was measured using an AR2000 Rheometer with parallel plate geometry.
- Dynamic tests at different ramp rates were performed to capture the changes in material viscosity as a function of temperature and degree of cure. A total of 8 tests at 1, 2, 3, and 4 C/min (two at each rate) were performed.
- Disposable aluminum plates of diameter 25mm were used at a gap of 1mm (sample thickness). Frequency of oscillation was chosen to be 1Hz.



## **Viscosity Measure**

Rheometry tests are performed under sinusoidal oscillatory loads. The classical solution of the viscoelastic behaviour of materials in such circumstances is usually expressed in terms of storage and loss moduli. Considering the following strain function, the resulting stress would be:

Strain applied: $\gamma = \gamma_0 \sin \omega t$ Resulting stress: $\sigma = \sigma_0 \sin(\omega t + \delta)$ 

where  $\delta$  is the phase angle between the strain and the stress. Decomposing the stress into in-phase and out-of-phase components, we get:

$$\sigma = \gamma_0 (G' \sin \omega t + G'' \cos \omega t)$$

where G' is the in-phase (elastic or storage) modulus and G'' is the out-ofphase (viscous or loss) modulus.

## **Viscosity Measure**

Alternatively, a complex viscosity  $\eta^* {\rm can}$  be defined with "dynamic viscosity" (  $\eta$ 





### Raw Data







## **Mathematical Model**

The viscosity model chosen is in the following form<sup>1</sup>:

$$\mu = \mu_1(T) + \mu_2(T) \frac{x_g}{x_g - x}$$
 (A+Bx+Cx<sup>2</sup>)

where

$$\mu_i(T) = \mu_{0i} e^{\frac{E_i}{RT}} \qquad ior 2$$

<sup>1</sup>Khoun and Hubert, Processing Characterization of a RTM Carbon Epoxy System for Aeronautical Applications.





## **Mathematical Model**

The parameters of the viscosity model are determined by fitting to the experimental results.

 $E_1 = 81,908$ 





## Dynamic Tests – 1cpm-1









## Dynamic Tests – 3cpm-2







## Dynamic Tests – 4cpm-2







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### **Resin Viscosity**



# Heat Capacity (Cp) Model





## Nomenclature

*Cp* Heat Capacity (J/gC)





## **Cp Measurement**

• The DSC tests were temperature modulated to obtain







### Raw Data















## **Cp Model Basics**





## Cp Model

The parameters of the Cp model are determined by fitting to the experimental results. The model is formulated as shown below:

$$C_{p_{ij}} = s_{ij}T + c_{ij} \quad (i = r, g \text{ and } j = 0, \infty)$$

$$(=) + \sum_{\infty} C_{p_g} - C_{p_r}$$

$$C_{p} = C_{p_{r}} + \frac{1}{1 + e^{k[(T - T_{g}) - \Delta T_{c}]}}$$

#### <u>Glassy</u>

**Rubbery** 

#### **Other parameters**

k = 0.278

 $\Delta T_c = -1.5$ 

### Valid for temperature values between -70°C to 275°C.

172



 $s_{g0} = 0.003775$ 







## Dynamic Tests – 2cpm-2





## Dynamic Tests – 3cpm







## Dynamic Tests – 5cpm

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## Dynamic Tests – 6cpm




### Dynamic Tests – 8cpm





### Dynamic Tests – 9cpm

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### Dynamic Tests – 10cpm

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#### Isothermal Tests – 100C









#### Isothermal Tests – 110C





# Isothermal Tests – 120C



#### Ism0.0006 Tc55/WSrmal Tests -









#### Isothermal Tests – 180C-1

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#### Isothermal Tests – 180C-2









# Dynamic Tests – 1Cpm





# Dynamic Tests – 1Cpm









# Dynamic Tests – 2Cpm-1









# Dynamic Tests – 3Cpm





**201** 





# **Dynamic Tests – 4Cpm0.00.51.0**





204













# Dynamic Tests – 9Cpm





# Dynamic Tests – 9Cpm







# Dynamic Tests – 10Cpm






# Isothermal Tests – 110C











# Isothermal Tests – 120C







#### Isothermal Tests – 130C





#### Isothermal Tests – 140C





### Isothermal Tests – 150C





# Isothermal Tests – 160C











# Isothermal Tests – 170C







#### Etalermal Tests - 180C-1012015







# Isothermal Tests – 180C-1











# Isothermal Tests – 190C







#### Material Model Verification – 100C

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#### Material Model Verification – 1Cpm

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